

# Recitation 1 9/8/04

## Comparison of viscosities

Great Molasses Flood (need website, photos) book reference, link

Wall of molasses 15-30 ft high  
traveling at 25-35 mph

Rum + baked beans

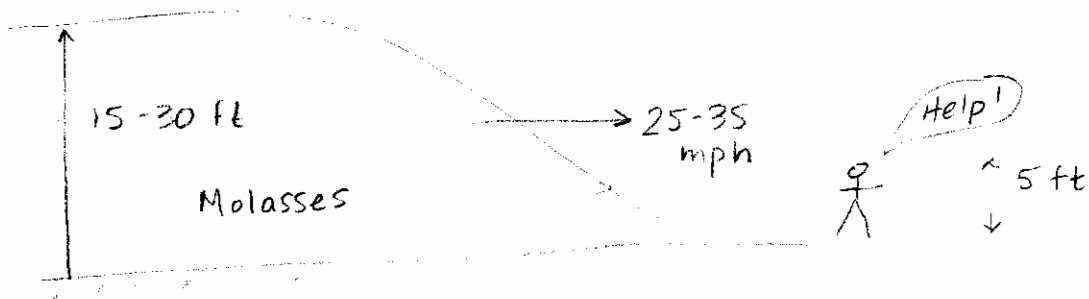
High molasses mark, smell in North end

50 ft tank, 2.2 million gallons

→ 20°F → 40°F increase  
full tank, prohibition  
factor of safety

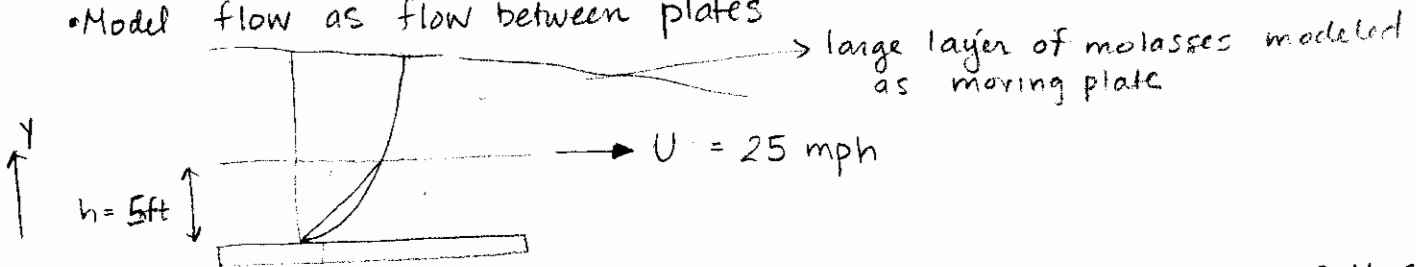
Shearing stress in molasses wave? at average height of person?

Sketch:



Assumptions:

• Model flow as flow between plates



- Assume linear velocity profile in bottom layer
- Laminar, newtonian fluid

Outline

- Problem Set-up
- Starch example
- Surface tension example

Property values:

$$\mu = 7500 \text{ cP}$$

$$1 \text{ cP} = 6.7197 \times 10^{-4} \text{ lbm/s.ft}$$

$$1 \text{ cP} = 3.6 \text{ Kg/hr.m}$$

Solution:

$$\tau = \mu \frac{du}{dy}$$

$$u = U \frac{y}{h}$$

$$\tau = \mu \frac{U}{h} = (1500) (6.7197 \times 10^{-4} \frac{\text{lbm}}{\text{s ft}}) \frac{(25 \frac{\text{mi}}{\text{hr}}) (5280 \frac{\text{ft}}{\text{mi}})}{(6 \text{ ft})} (\frac{1 \text{ hr}}{3600 \text{ s}})$$

$$\tau = \frac{30.7 \frac{\text{lbm}}{\text{ft s}^2}}{32.174 \frac{\text{lbm ft}}{\text{lb f s}^2}} = 0.95 \frac{\text{lb f}}{\text{ft}^2} \left(\frac{1 \text{ ft}}{12 \text{ in}}\right)^2 = .006 \text{ psi}$$

What caused damage?

Pressure in flowing fluid

Stickiness of molasses

Other fluids?

Review of problem solving methodology in fluids:

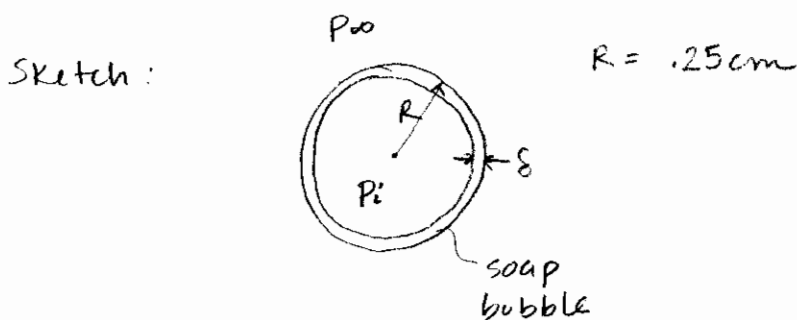
- Sketch or figure - label important features
- Assumptions - know what these are!
- Approach
- Property values - look up what you will need first + convert units if necessary
- Solution - be neat, line by line, waste paper
- Reflection - does the answer make sense? can you compare it to anything else?

Overall: 1 problem per page, ink or pencil fine

## Surface tension example problem

3

Problem: What is the pressure difference between the inside and outside of 0.5 cm diameter soap bubble?

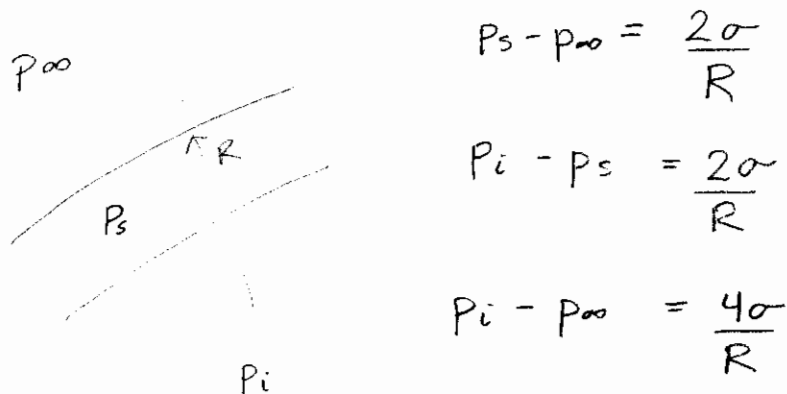


Assumptions:  $\delta \ll R$   
gravity is negligible

Approach: Surface tension force balance

Properties:  $\sigma_{\text{soap}} = \frac{1}{3} \sigma_{\text{H}_2\text{O}} = \frac{1}{3} (.07 \text{ N/m}) = .023 \text{ N/m}$   
 $\rho_{\text{soap}} \approx \rho_{\text{water}} = 1000 \text{ kg/m}^3$

Solution:



$$P_s - P_o = \frac{2\sigma}{R}$$

$$P_i - P_s = \frac{2\sigma}{R}$$

$$P_i - P_o = \frac{4\sigma}{R}$$

$$\Delta p = \frac{4(.023 \text{ N/m})}{(.25 \text{ cm}) \left( \frac{1 \text{ m}}{100 \text{ cm}} \right)} = 2.3 \text{ Pa}$$

Reflections: Soap has less surface tension than  $\text{H}_2\text{O}$  → why is it harder to blow a water bubble?

[website]